

Aromatic Lipophilic Spin Traps Effectively Inhibit RPE65 Isomerohydrolase Activity

Eugenia Poliakov,[†] Toral Parikh,[†] Michael Ayele, Stephanie Kuo, Preethi Chander, Susan Gentleman, and T. Michael Redmond*

LRCMB, NEI, National Institutes of Health, Bethesda, Maryland 20892, United States

Supporting Information

ABSTRACT: We previously showed that RPE65 does not specifically produce 11-cis-retinol only but also 13-cis-retinol, supporting a carbocation or radical cation mechanism of isomerization. The intrinsic properties of conjugated polyene chains result in facile formation of radical cations in oxidative conditions. We hypothesized that such radical intermediates, if involved in the mechanism of RPE65, could be stabilized by spin traps. We tested a variety of hydrophilic and lipophilic spin traps for their ability to inhibit RPE65 isomerohydrolase activity. We found that the aromatic lipophilic spin traps such as *N-tert*-butyl-α-phenylnitrone (PBN), 2,2-dimethyl-4-phenyl-2*H*-imidazole-1-oxide (DMPIO), and nitrosobenzene (NB) strongly inhibit RPE65 isomerohydrolase activity *in vitro*.

The retinyl ester isomerohydrolase central to the visual cycle was recently confirmed biochemically to be RPE65, 1-3 a protein highly preferentially expressed in the RPE.⁴ Prior molecular genetics findings, in a targeted mouse model⁵ and in human RPE65 gene mutations, ^{6–9} suggested a key role for RPE65 in isomerization. RPE65 is a member of a family of carotenoid oxygenases that specialize in oxidative cleavage of double bonds in various carotenoids. However, RPE65 itself does not perform oxidative cleavage. Rather, by concerted O-alkyl cleavage of all-trans-retinyl ester and isomerization of the retinyl moiety, it produces 11-cis-retinol and a fatty acid. 10 We recently presented evidence in support of the carbocation or radical cation mechanism by showing that RPE65 is capable of enzymatic isomerization to 13-cis-retinol.¹¹ Indeed, we found that site-directed mutagenesis of a single residue, Phe103, converted the mutant RPE65 into a preferential 13-cis-retinol isomerohydrolase, and our data were supported by identification of specific 13-cis-retinol RPE65 isomerohydrolase in zebrafish. 12 In light of this, we concluded that retinoid metabolism in the visual cycle occurs by a carbocation mechanism (S_N1 hydrolysis of retinyl ester)¹³ or by a cation radical mechanism (one electron oxidation of retinyl ester)11 as the data could be explained either way. Carotenoid radical cations have a tendency to lose a proton to form a neutral carotenoid radical.¹⁴ If the latter mechanism were true, we conjectured that RPE65 could be inhibited by spin trap compounds.

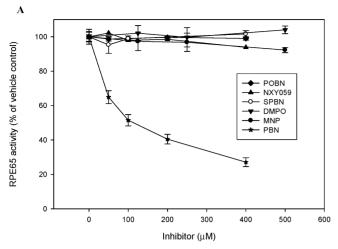
To test this hypothesis, we chose a variety of cell permeable spin traps and tested these in a robust minimal visual cycle (canine RPE65/bovine LRAT) reconstructed in HEK293F cells.³ HPLC separation of retinol isomers was conducted as previously described.³ RPE65 isomerohydrolase activity was calculated as ratio of 11-cis-retinol or 13-cis-retinol to a total amount of all retinol isomers (11-cis-, 13-cis-, and all-transretinol). We discovered that several lipophilic aromatic spin traps are effective inhibitors of RPE65 isomerase activity. We confirmed our findings using a bovine RPE microsomal fraction preparation.

Various spin traps compounds including PBN, 2-methyl-2nitrosopropane (MNP), and nitrosobenzene have been successfully employed to stabilize and detect carotenoid and retinoid radicals by EPR spectroscopy. 14,15 In our cell-based assay both PBN and nitrosobenzene demonstrated inhibitory effect at 100 μ M concentrations (52 \pm 3% and 32 \pm 5% of vehicle control RPE65 activity, respectively; Table S1). 5,5-Dimethyl-1-pyrroline-N-oxide (DMPO) does not form stable adducts with carotenoid radicals in ESR experiments 14 and predictably did not have an inhibitory effect on RPE65 activity (Figure 1). Nitrosobenzene has a limited usefulness because of its well-documented toxicity in cells and animals. 16,17 On the other hand, PBN is a widely used agent that is well tolerated in animals, while a 1,3-disulfonate derivative of PBN, disufenton sodium (NXY059), has been tested in humans as a neuroprotective agent for stroke. 18,19 We found that PBN is an effective uncompetitive inhibitor of RPE65 with $K_i = 61 \mu M_i$ which is in the same range as its effects on oxygen radicals (Figures 1 and 2). RPE65 protein expression is not reduced at these concentrations, nor is total retinyl ester accumulation, indicating a direct inhibition of RPE65 (data not shown). PBN instability in vivo has been reported, and its neuroprotective effects have been attributed to the degradation products tertbutylhydroxylamine (TBH) and consequently MNP.²⁰ Neither of these compounds had an effect on RPE65 activity in our cellbased assay at 200 μ M. DMPIO, a structural and functional analogue of PBN, demonstrated a significant inhibitory potency while a structural analogue that lacks an N-oxide moiety, nbenzylidene-tert-butylamine (NBTB), demonstrated much less inhibitory effect on RPE65 activity. Moreover, spin traps inhibit 13-cis isomer formation to the same extent as 11-cis isomer (Table 1). The uncompetitive nature of PBN inhibition was shown by the lowering of IC₅₀ as substrate concentration was increased (Figure 2). These findings are consistent with the proposed mechanism of trapping of retinyl ester radical

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Biochemistry Rapid Report



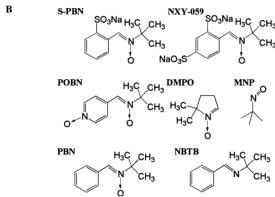


Figure 1. RPE65 activity in the presence of various spin traps. (A) HEK293F cells transfected as described (see Supporting Information) were incubated in the presence of 0–500 μ M of POBN, NXY059, SPBN, DMPO, MNP, and PBN at 2.5 μ M *all-trans*-retinol. Activity was calculated relative to the vehicle control; $n \geq 3$ for each data point. (B) Structures of spin traps and a structural analogue used in this work.

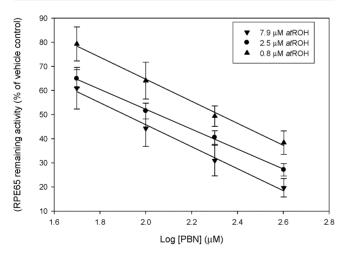


Figure 2. Inhibition of RPE65 activity in the presence of PBN at various substrate concentrations (IC₅₀ determination). HEK293F cells transfected as described (see Supporting Information) were incubated in the presence of 0–400 μ M PBN in DMSO (vehicle) at substrate concentrations of 0.79, 2.5, or 7.9 μ M *all-trans*-retinol (half-log concentration steps). Activity was calculated relative to DMSO vehicle control. Data plotted as regression against log PBN concentration; $n \ge 4$ for each data point.

Table 1. RPE65 Activity (%) in the Presence of PBN, Possible Degradation Products, a Spin Trap Analogue DMPIO, and Structural Analogue NBTB without Spin Trap Properties at 2.5 μ M Substrate

inhibitor (200 μ M)	RPE65 activity (11-cis isomer % of vehicle)	RPE65 activity (13-cis isomer % of vehicle)
PBN	50 ± 6	43 ± 6
TBH	103 ± 6	107 ± 10
MNP	94 ± 5	95 ± 7
NBTB	86 ± 6	91 ± 6
DMPIO	68 ± 6	63 ± 1

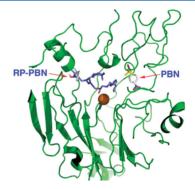


Figure 3. Simulated docking of PBN—retinyl palmitate spin adduct on RPE65 model. Models of a spin adduct of *all-trans*-retinyl palmitate and PBN (RP-PBN; purple) and PBN (yellow) were docked independently on a composite RPE65 model as described (see Supporting Information), but depicted here together.

intermediate by PBN. We modeled RPE65 on a composite structural template (VP14, ACO, and RPE65; see Supporting Information) to ensure an "open" conformation of the protein. Docking of a PBN-retinyl palmitate spin adduct was performed using AutodockVina and compared with docking of the PBN molecule alone (Figure 3), predicting that while the spin adduct could dock favorably (–9.5 kcal/mol), PBN alone could not (–5.6 kcal/mol), remaining distant from the iron center, ruling out the possibility of iron chelation, and so not expected to directly affect the catalytic activity. Modeling attempts with the RPE65 structure alone were unsuccessful due to the "closed" substrate binding cleft of the RPE65 crystal in the absence of substrate²¹ and possibly in the presence of reaction products [Chander, P., manuscript in preparation].

In considering more hydrophilic spin traps, we analyzed the effect on RPE65 activity of *N-tert*-butyl-(2-sulfophenyl)nitrone (SPBN), α -(4-pyridyl 1-oxide)-*N-tert*-butylnitrone (POBN), and NXY059. However, none of these inhibited RPE65 activity in our cell-based assay up to 400 μ M (Figure 1). Therefore, we concluded that spin trap hydrophobicity is a key element to access the substrate tunnel and to bind to a reaction radical intermediate.

However, when we treated cells with a number of unrelated hydrophobic spin traps, only the aromatic N-oxides DMPIO and 2-(2-carboxyethyl)-2-methyl-4-phenyl-2H-imidazole-1-oxide (MCPIO) demonstrated a moderate inhibitory effect at 100 μ M concentration (Table S1). There was no substantial change in cell viability (\sim 90%) and RPE65 protein expression in treated cells (Table S2). We confirmed in the isolated HEK293 microsomal fraction that PBN and DMPIO significantly inhibit RPE65 activity at 200 μ M inhibitor (46%)

Biochemistry Rapid Report

and 56% of vehicle, respectively). Also, we demonstrated that production of 11-cis-retinol from bovine RPE microsomal fraction is inhibited by PBN at 400 μ M (53 \pm 3% of wild-type RPE65 activity).

Thus, we describe for the first time a new class of RPE65 inhibitors that are based on trapping of a reaction intermediate with hydrophobic aromatic N-oxides. Inhibitory potency decreases with increase in polarity from NB to PBN to DMPIO and to MCPIO (Table S1). On the other hand, NBTB, a structural analogue of PBN without spin trap properties does not inhibit RPE65 activity significantly, and thus we conclude that the N-oxide group is involved in RPE65 inhibition. The inhibition of RPE65 activity is a possible route for slowing the visual cycle and preventing accumulation of A2E byproducts involved in the etiology of Stargardt macular dystrophy. Interestingly, PBN protects rats against light damage at 50 mg/kg systemic administration, 22 an effect ascribed to scavenging of oxygen radicals due to rhodopsin photoproducts. However, our data suggest that this protective effect may be upstream, at least in part, due to inhibition of RPE65 activity. Reduced 11-cis-retinoid flux via inhibition of RPE65 would reduce rhodopsin photocycling and so reduce generation of free radicals caused by light damage. This is pertinent as $Rpe65^{-/-}$ mice and C57BL/6 mice with the hypomorphic L450 M RPE65 variant are both much less susceptible to light damage than wild-type mice.²³ We conclude that inhibition by spin traps of RPE65 lends strong support to a radical cationbased mechanism of retinol isomerization by RPE65.

ASSOCIATED CONTENT

Supporting Information

Detailed experimental methods and additional tables. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*Tel: 301-496-0439. Fax: 301-402-1883. E-mail: redmond@helix.nih.gov.

Author Contributions

[†]Authors contributed equally toward this publication.

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